#### HETEROCYCLES, Vol. 65, No. 6, 2005, pp. 1439 - 1445 Received, 2nd February, 2005, Accepted, 23rd March, 2005, Published online, 24th March, 2005

# 1,4,5,6-TETRAHYDROPYRIMIDINIUM HALIDES LIGANDS FOR SUZUKI-MIYAURA CROSS-COUPLING OF UNACTIVATED ARYL CHLORIDES

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**Abstract-** Four functionalized bis(1,4,5,6-tetrahyropyrimidinium) salts (**2a-d**) have been prepared and characterized by conventional spectroscopic methods and elemental analyses. A highly effective, easy to handle and environmentally benign process for palladium mediated Suzuki cross-coupling was developed. The *in situ* prepared three component system  $Pd(OAc)_2$  / bis(1,4,5,6-tetrahyropyrimidinium) bromides (**2a-d**) and  $Cs_2CO_3$  catalyses quantitatively the Suzuki cross-coupling of deactivated aryl chloride.

## **INTRODUCTION**

Suzuki coupling reaction<sup>1</sup> has, over the past decade or so, established itself as a powerful method for the formation of carbon-carbon bonds. The coupling of two sp<sup>2</sup>-centered carbons remains an important goal as well as a challenge for synthetic organic chemists. Chloroarenes are generally readily available and inexpensive and, therefore the productive activation of C-Cl bonds in chloroarenes is of significant industrial interest. Traditionally, however, the C-Cl bond in chloroarenes has been found to be comparatively inert.<sup>2</sup> For along time, the use of unactivated aryl chlorides as substrates in Suzuki-Miyaura cross-coupling was not feasible. Instead, more costly and less readily available aryl iodides and aryl bromides had to be used. Buchwald<sup>3</sup> and Fu<sup>4</sup> were the first to independtly developed catalyst systems based on electron-rich, sterically demanding phosphine ligands, which allow the Suzuki-Miyaura cross-coupling of many unactivated aryl chlorides. However, the major drawback of these is that the phosphine ligands are comparatively difficult to make or rather expensive. Furthermore, tertiary phosphines require air-free handling to prevent their oxidation and are susceptible to P-C bond cleavage at elevated temperatures.<sup>5</sup> The application of *N*-heterocyclic carbene (NHC)<sup>6</sup> ligands in the reaction was first reported by Herrmann in 1998.<sup>7</sup>

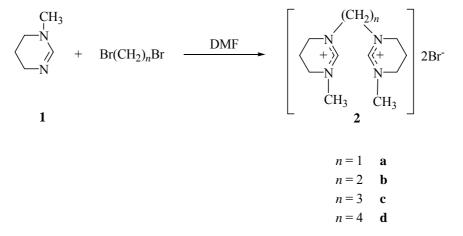
The NHC complexes are cost efficient to prepare, insensitive to air and moisture and are thermally stable in both the solid state and in solution; the carbenes are non-dissociative ligands. However, the development of new ligands or the application of existing ligands in these reactions, particularly those involving aryl chlorides

as substrates, is still of considerable importance. Recently, we have developed improved procedures Heck and Suzuki reactions of aryl chlorides making use of novel ligands 1,3-bis(dialkyl)imidazolium and benzimidazolinium salts.<sup>8</sup>

Although the nature of the NHC ligand on complexes has a tremendous influence on the rate of catalyzed reactions, the use of 1,4,5,6-tetrahyropyrimidinium ligands in coupling reactions is a neglected area. In order to find more efficient palladium catalysts we have prepared a series of new bis(1,4,5,6-tetrahyropyrimidinium) bromides LHX (2) (Scheme 1). We now report the use of the *in situ* generated catalytic system composed of commercially available and stable reagents, the Pd(OAc)<sub>2</sub> as palladium source, bis(1,4,5,6-tetrahyropyrimidinium) (**2a-d**) as a carbene precursor and  $Cs_2CO_3$  as a base for cross coupling of aryl chlorides with phenylboronic acid.

### **RESULTS AND DISCUSSION**

Bis(1,4,5,6-tetrahyropyrimidinium) bromides (**2a-d**) are conventional NHC precursors. According to Scheme 1, the salts (**2a-d**) were obtained in almost quantitative yield by quarternazition of 1-alkyl(1,4,5,6-tetrahyropyrimidine)<sup>9</sup> in DMF with alkyl bromides.<sup>8</sup>



#### Scheme 1

The structures of **2** were determined by their characteristic spectroscopic data and elemental analyses (EXPERIMENTAL). <sup>13</sup>C NMR chemical shifts were consistent with the proposed structure; the imino carbon appeared as a typical singlet in the <sup>1</sup>H-decoupled mode in the 154.8, 154.4, 150.5 and 153.5 ppm respectively for pyrimidinium salts **2a-d**. The <sup>1</sup>H NMR spectra of the pyrimidinium salts furher supported the assigned structures; the resonances for C(2)-H were observed as sharp singlets in the 8.7, 8.5, 8.4 and 8.3 ppm respectively for **2a-d**. The IR spectral data for pyrimidinium salts (**2a-d**) clearly indicate the presence of the – C=N- group with a v(C=N) vibration at 1697, 1694, 1690 and 1699 cm<sup>-1</sup> respectively for **2a-d**.

The palladium-catalyzed cross-coupling of arylboronic acids with aryl halides has been shown to proceed under a variety of conditions: A wide range of bases and solvents, as well as catalysts, have been employed with varying degrees of success according to the substrates. To find optimum conditions a series of experiments has been performed with 4-chloroanisole and phenylboronic acid as model compounds. As a base,  $Cs_2CO_3$  was the best choice and as a solvent dioxane was found to be better than other solvents. After having established the optimised coupling reaction conditions, the scope of the reaction and efficiencies of the salts were evaluated by investigating the coupling of  $C_6H_5B(OH)_2$  with various *p*-substituted aryl chlorides. The results were summarized in Table 1.

Table 1. The	Suzuki coupli	ng reaction	of arvl	chlorides	with phe	enylboronic acid	ł
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$R \longrightarrow Cl + $	}—B(OH) <sub>2</sub>	$\frac{Pd(OAc)_2 (1.0 \text{ mol }\%)}{2a-d (2.0 \text{ mol }\%)}$		) $R \rightarrow R$
	Entry	R	LHX	Yield <sup>a-d</sup> (%)
	1	COCH <sub>3</sub>	2a	93
	2	COCH <sub>3</sub>	<b>2b</b>	95
	3	COCH <sub>3</sub>	<b>2c</b>	89
	4	COCH <sub>3</sub>	2d	87
	5	CH <sub>3</sub>	2a	80
	6	CH <sub>3</sub>	2b	84
	7	$CH_3$	2c	76
	8	CH <sub>3</sub>	2d	73
	9	СНО	2a	94
	10	СНО	<b>2b</b>	96
	11	СНО	2c	91
	12	СНО	<b>2d</b>	89
	13	OCH <sub>3</sub>	2a	80
	14	OCH <sub>3</sub>	2b	86
	15	OCH <sub>3</sub>	2c	84
	16	OCH <sub>3</sub>	2d	81
	17	Н	2a	90
	18	Н	2b	93
	19	Н	2c	86
	20	Н	2d	82

<sup>a</sup> *Reaction conditions*: 1.0 mmol of R-C<sub>6</sub>H<sub>4</sub>Cl-*p*, 1.5 mmol of phenylboronic acid, 2 mmol Cs<sub>2</sub>CO<sub>3</sub>, 1 mmol % Pd(OAc)<sub>2</sub>, 2 mmol % **2**, dioxane (3 mL), <sup>b</sup> Purity of compounds is checked by NMR spectrum and yields are based on aryl chloride. <sup>c</sup> All reactions were monitered by TLC, <sup>d</sup> temperature 80 °C, 2 h.

Under those conditions, *p*-chloroacetophenone, *p*-chlorotoluene, *p*-chlorobenzaldehyde, *p*-chloroanisole and chlorobenzene react very cleanly with phenylboronic acid in goods yields (Table 1, Entries 2, 6, 10, 14 and 18).

Table 1 summarizes our results from the screening of four pyrimidinium salts, for Suzuki cross-coupling reaction. Several trends are readily apparent. First, the procedure is simple and does not require induction periods. Second, the scope of this reaction is broad and includes aryl chlorides that are activated or

deactivated. Third, all complexes led to good conversions at low catalyst concentration (1 mmol %). It is evident that the NHC precursors that contain pyrimidinium (2a-d) are the most effective of the salts examined.

In conclusion, we have developed a new type of easily prepared bis(1,4,5,6-tetrahyropyrimidinium) salts LHX (**2a-d**) ligands in the Suzuki-Miyaura coupling reaction. Through the use of LHX and Pd(OAc)<sub>2</sub> as a precatalyst mixture, aryl halides undergo efficient coupling reactions with C<sub>6</sub>H<sub>5</sub>B(OH) <sub>2</sub> in the presence of Cs<sub>2</sub>CO<sub>3</sub>. The procedure is simple and efficient towards various aryl halides and does not require induction periods.

### **EXPERIMENTAL**

All reactions were performed using Schlenk-type flask under argon and standard high vacuum-line techniques. Solvents were analytical grade and distilled under Ar from sodium benzophenone (Et<sub>2</sub>O, dioxane). <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded using a Bruker AC300P FT spectrometer operating at 300.13 MHz (<sup>1</sup>H), 75.47 MHz (<sup>13</sup>C). FT-IR spectra were recorded on a Mattson 1000 spectrophotometer. Elemental analyses were performed by TUBITAK Microlab.

#### Preparation of 3,3'-dimethyl-1,1'-methylenedi(1,4,5,6-tetrahyropyrimidinium) diibromide (2a)

To a solution of 1-methyl-1,4,5,6-tetrahydropyrimidine ( 0.98 g, 10 mmol) in DMF (10 mL) was added slowly dibromomethane (0.87 g, 5 mmol) at 25 °C and the resulting mixture was stirred at rt for 6 h. Ether (15 mL) was added to obtain a white crystalline solid which was filtered off. The solid was washed with ether (3x15 mL), dried under vacuum and the crude product was recrystallized from ethanole / ethere. Yield: 1.65 g, 89 %. mp 197-198 °C, IR (KBr)  $\upsilon$ = 1697 cm<sup>-1</sup> (-CH=N-). <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, )  $\delta$  : 2.2 (4H, quintet, J = 5.9 Hz, H-5), 3.3 (6H, s, NCH<sub>3</sub>); 3.5 (4H, t, J = 6.0 Hz, H-6)], 3.6 (4H, t, J = 5.8 Hz, H-4); 5.2 (2H, s, H-bridge); 8.7 (2H, s, 2-H). <sup>13</sup>C NMR (DMSO-d<sub>6</sub>)  $\delta$  : 18.5 (NCH<sub>3</sub>); 42.3 (C-5), 45.6 (C-6), 46.8 (C-4), 71.2 (C-bridge), 154.8 (C-2). Anal. Calcd for C<sub>11</sub>H<sub>22</sub>N<sub>4</sub>Br<sub>2</sub>: C, 35.70; H, 5.99; N, 15.14. Found: C, 35.66; H, 5.89; N 15.11.

#### Preparation of 3,3'-dimethyl-1,1'-ethylenedi(1,4,5,6-tetrahyropyrimidinium) dibromide (2b)

Compound (**2b**) was prepared in the same way as **2a** from 1-methyl-1,4,5,6-tetrahydropyrimidine (0.98 g, 10 mmol) and 1,2-dibromoethane (0.94 g, 5 mmol) to give white crystals of **2b**. The crude product was recrystallized from ethanole / ethere. Yield: 1.75 g, 91 %. mp 242-243 °C, IR  $\upsilon$  = 1696 cm<sup>-1</sup> (-CH=N-). <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  : 2.2 (4H, quintet, *J*= 5.8 Hz, H-5), 3.3 (6H, s, NCH<sub>3</sub>), 3.5 (4H, t, *J* = 6.0 Hz, H-6), 3.6 (4H, t, *J*= 5.9 Hz, H-4), 3.9 (4H, s, NCH<sub>2</sub>CH<sub>2</sub>N), 8.5 (2H, s, H-2). <sup>13</sup>C NMR (DMSO-d<sub>6</sub>)  $\delta$  : 18.9 (NCH<sub>3</sub>),

### Preparation of 3,3'-dimethyl-1,1'-propylenedi(1,4,5,6-tetrahyropyrimidinium) dibromide (2c)

Compound (**2c**) was prepared in the same way as **2a** from 1-methyl-1,4,5,6-tetrahydropyrimidine ( 0.98 g, 10 mmol) and 1,3-dibromopropane (1.01 g, 5 mmol) to give white crystals of **2c**. The crude product was recrystallized from ethanole / ethere. Yield: 1.80 g, 91 %. mp 120-121°C, IR  $\upsilon$ = 1690 cm<sup>-1</sup> (-CH=N-). <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$  : 2.2 (4H, quintet, *J* = 6.1 Hz, H-5); 3.3 (6H, s, NCH<sub>3</sub>); 3.5 (12H, m, H-6 and bridge NCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>N ), 3.6 (4H, t, *J*= 7.3 Hz, H-4), 8.4 (2H, s, H-2). <sup>13</sup>C NMR (DMSO-d<sub>6</sub>)  $\delta$ : 8.9 (NCH<sub>3</sub>), 25.8 (bridge, -NCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>N=), 41.4 (C-5), 42.68 (C-6), 44.9 (C-4), 52.03 (bridge - NCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>N-), 150.5 (C-2). Anal. Calcd for C<sub>13</sub>H<sub>26</sub>N<sub>4</sub>Br<sub>2</sub>: C, 39.21; H, 6.58; N, 14.07. Found: C, 39.15; H, 6.51; N, 13.98.

## Preparation of 3,3'-dimethyl-1,1'-buthylenedi(1,4,5,6-tetrahyropyrimidinium) dibromide (2d)

Compound (**2d**) was prepared in the same way as **2a** from 1-methyl-1,4,5,6-tetrahydropyrimidine ( 0.98 g, 10 mmol) and 1,4-dibromobutane ( 1.08 g, 5 mmol) to give white crystals of **2d**. The crude product was recrystallized from ethanole / ethere. Yield: 1.92 g, 93 %. mp 138-139°C, IR  $\upsilon$ = 1699 cm<sup>-1</sup> (-CH=N-). <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$ : 1.7-1.8 (4H, m, bridge -NCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>N), 2.2 (4H, quintet, *J* = 5.8 Hz, H-5), 3.3 (6H, s, NCH<sub>3</sub>), 3.3-3.4 (12H, m, ring H-4,-6 and bridge-NCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>N), 8.3 (2H, s, H-2). <sup>13</sup>C NMR (DMSO-d<sub>6</sub>)  $\delta$  : 18.9 (NCH<sub>3</sub>), 24.5 (bridge -NCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>N-), 41.4 (C-5), 42.6 (C-6), 44.9 (C-4), 54.6 (bridge - NCH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>N-), 153.5 (C-2). Anal. Calcd for C<sub>14</sub>H<sub>28</sub>N<sub>4</sub>Br<sub>2</sub>: C, 40.79; H, 6.85; N, 13.59. Found: C, 40.64; H, 6.49; N, 13.89.

General Procedure for the Suzuki Type Coupling reactions.  $Pd(OAc)_2$  (1 mmol %), bis(tetrahydropyrimidinium) bromides (2a-d) (2 mmol %), aryl chloride (1.0 mmol), phenylboronic acid (122 mg; 1.5 mmol), Cs<sub>2</sub>CO<sub>3</sub> (652 mg; 2 mmol) dioxane (3 mL) were added in a small Schlenk tube under argon and the mixture was heated at 80 °C for 2 h. At the conclusion of the reaction mixture was cooled, extracted with ther, filtered through a pad of silicagel with copious washings, concentrated and purified by flash chromatography on silicagel. Purity of compounds is checked by NMR spectrum and yields are based on aryl chloride.

## ACKNOWLEDGEMENTS

The Technological and Scientific Research Council of Türkiye TÜBİTAK TBAG-2474 (104T085), (TÜBİTAK COST D17) and İnönü University Research Fund are gratefully acknowledged for support of this work.

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